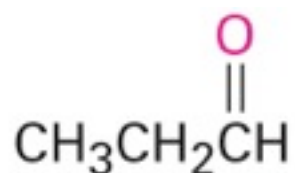
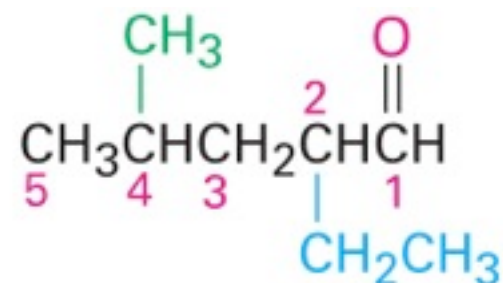


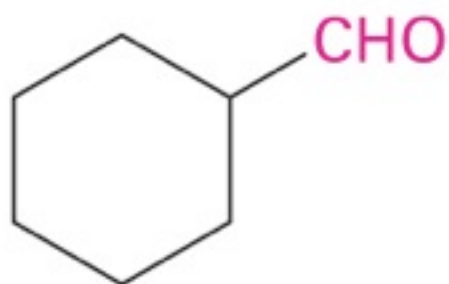
Ethanal
(acetaldehyde)



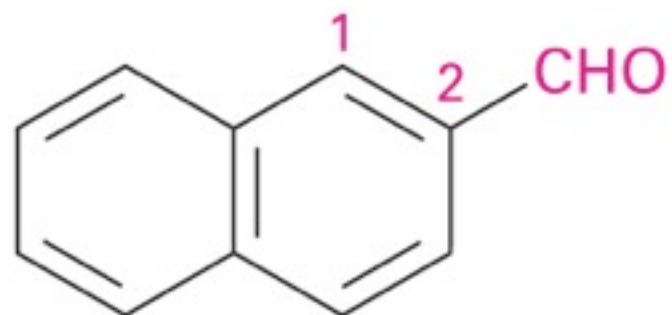
Propanal
(propionaldehyde)



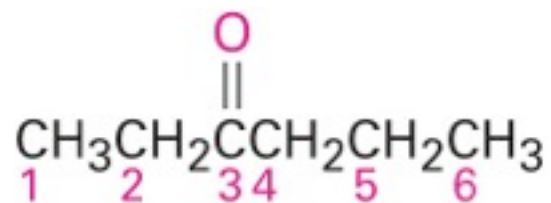
2-Ethyl-4-methylpentanal



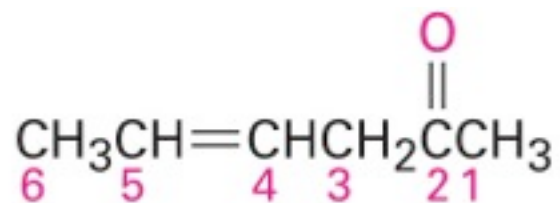
Cyclohexanecarbaldehyde



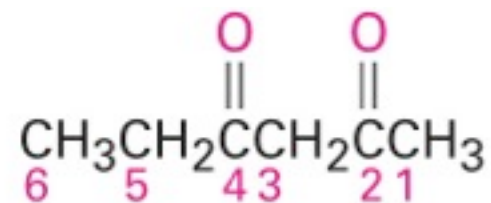
2-Naphthalenecarbaldehyde



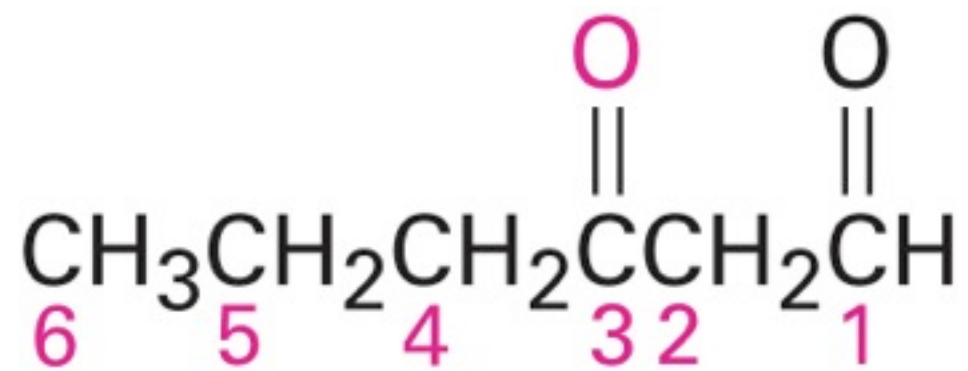
3-Hexanone
(New: Hexan-3-one)



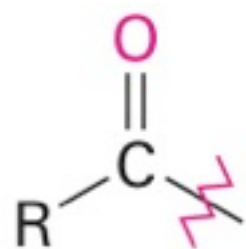
4-Hexen-2-one
(New: Hex-4-en-2-one)



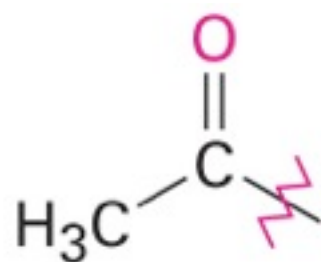
2,4-Hexanedione
(New: Hexane-2,4-dione)



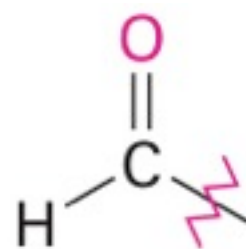
3-Oxohexanal



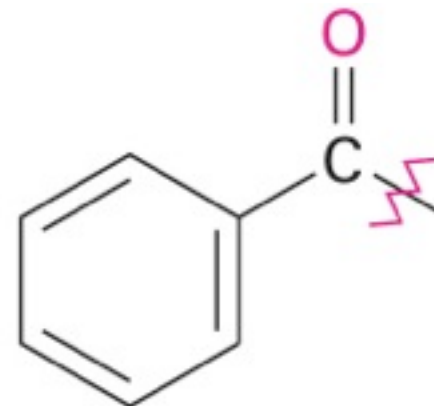
An acyl group



Acetyl

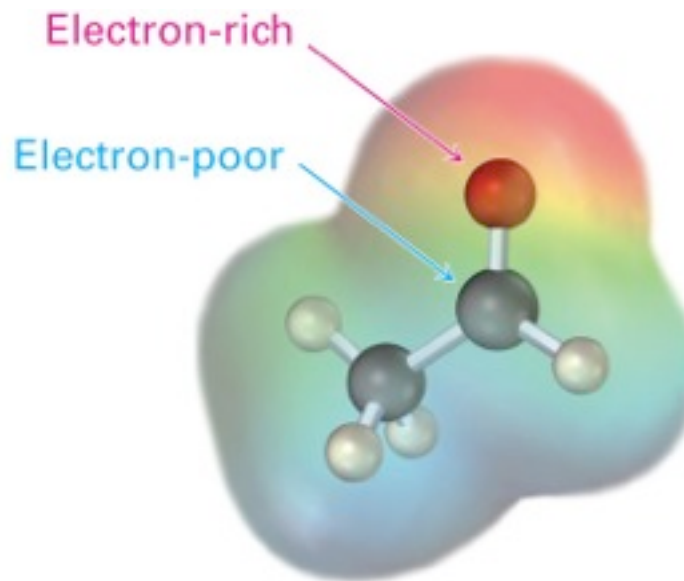


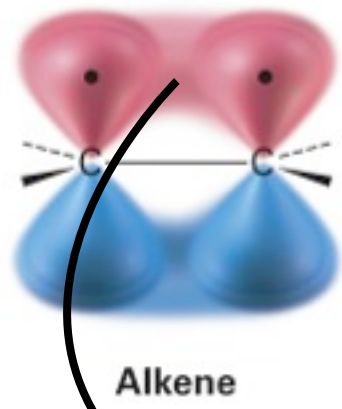
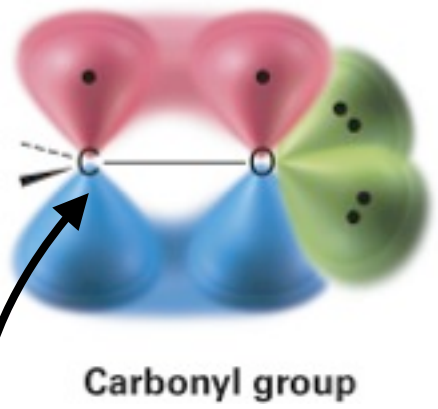
Formyl



Benzoyl

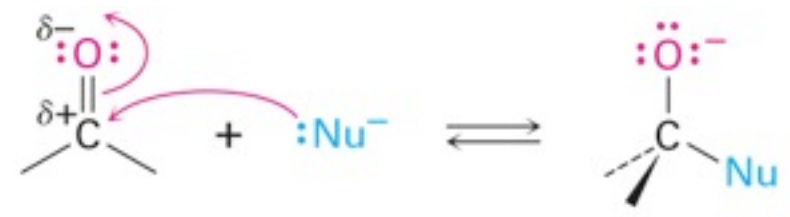
Preview of Carbonyl Chemistry





Nu:⁻

E⁺



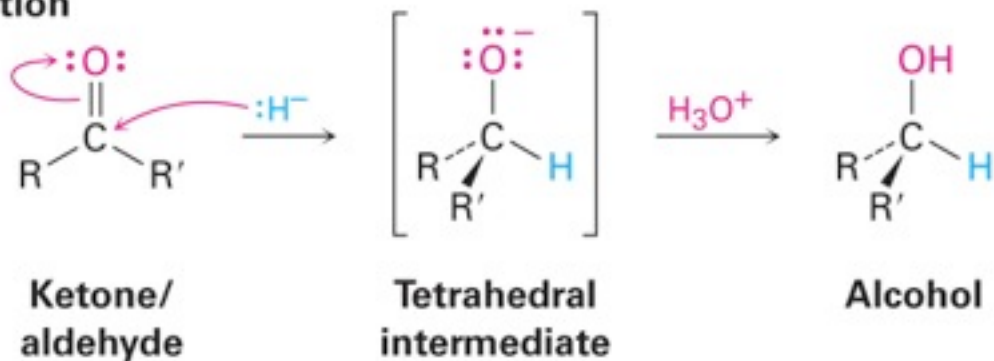
A carbonyl compound
(*sp*²-hybridized carbon)

A tetrahedral intermediate
(*sp*³-hybridized carbon)

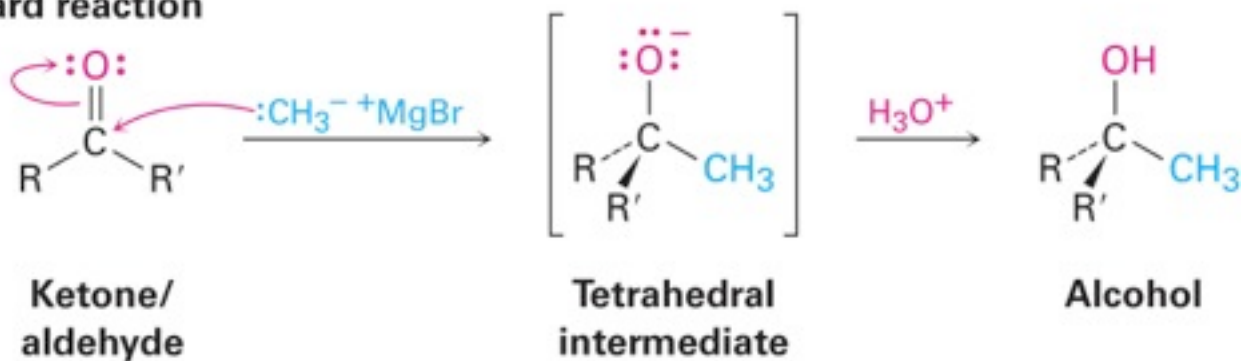
Nucleophilic
Acyl Addition

Nucleophilic Acyl Addition

Reduction

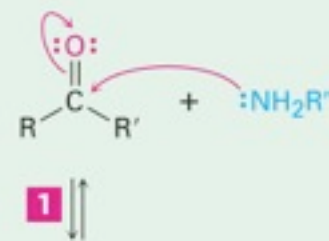


Grignard reaction

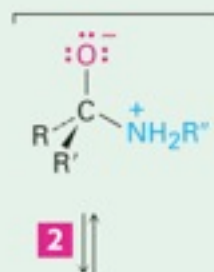


Nucleophilic Acyl Addition by Amines is Unique

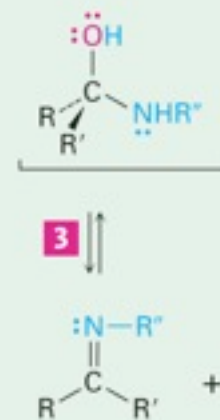
1 Addition to the ketone or aldehyde carbonyl group by the neutral amine nucleophile gives a dipolar tetrahedral intermediate.

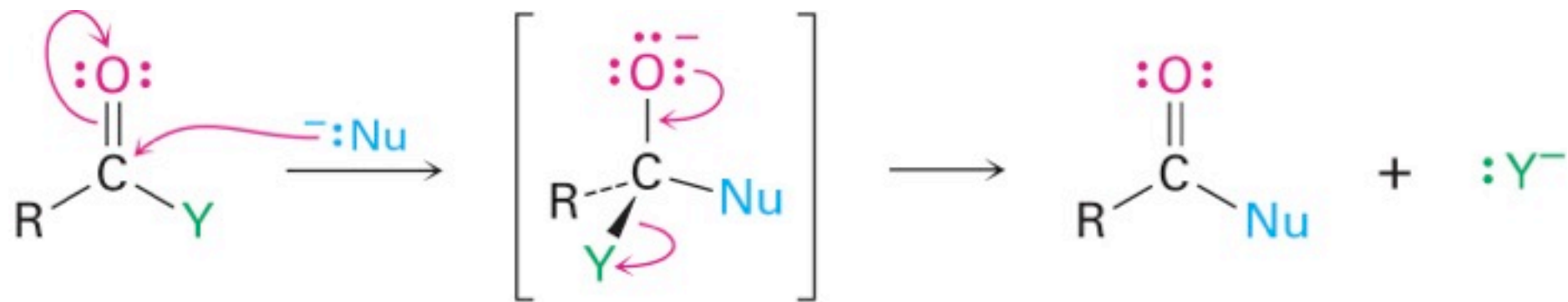


2 Transfer of a proton from nitrogen to oxygen then yields an amino alcohol intermediate.



3 Dehydration of the amino alcohol intermediate gives neutral imine plus water as final products.





**Carboxylic acid
derivative**

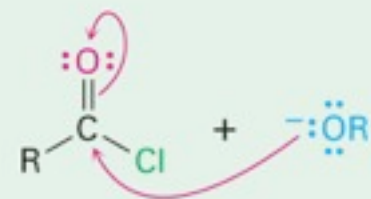
**Tetrahedral
intermediate**

**Nucleophilic Acyl
Substitution**

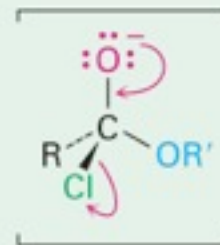
[$Y = -OR$ (ester), $-Cl$ (acid chloride),
or $-OCOR$ (acid anhydride)]

Nucleophilic Acyl Substitution

1 Nucleophilic addition of alkoxide ion to an acid chloride yields a tetrahedral intermediate.

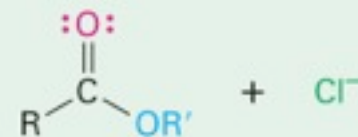


1 ↓

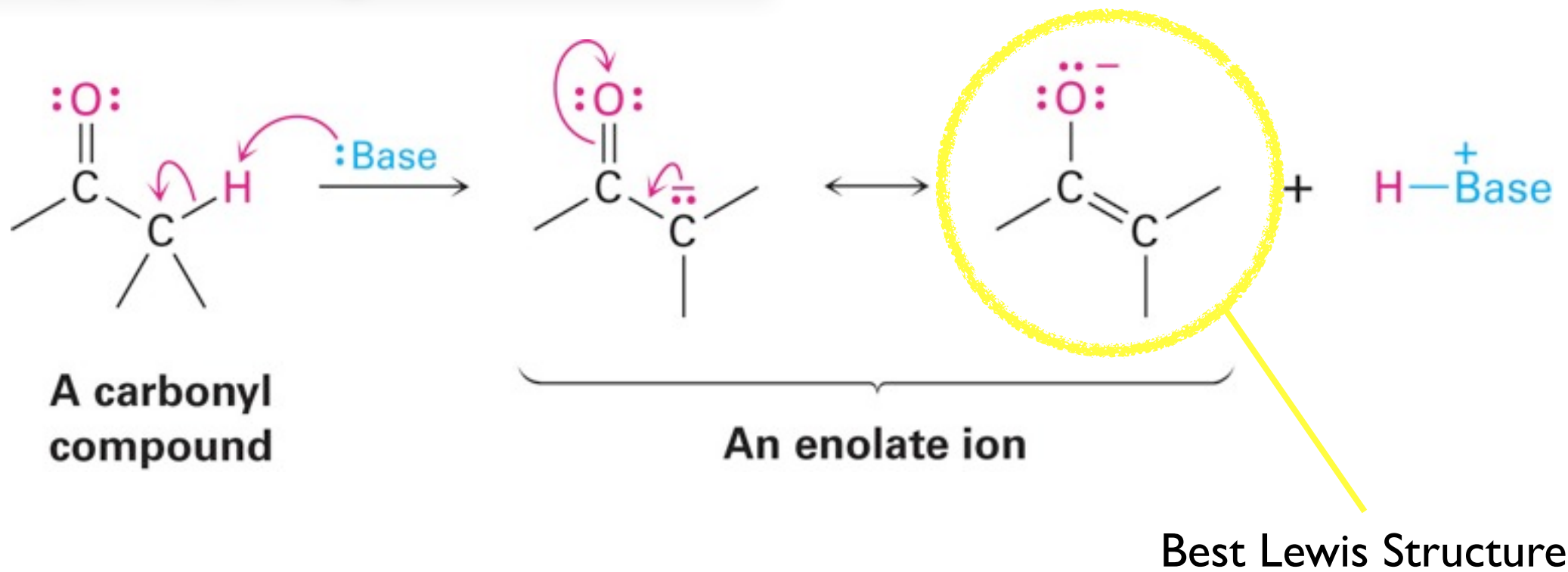


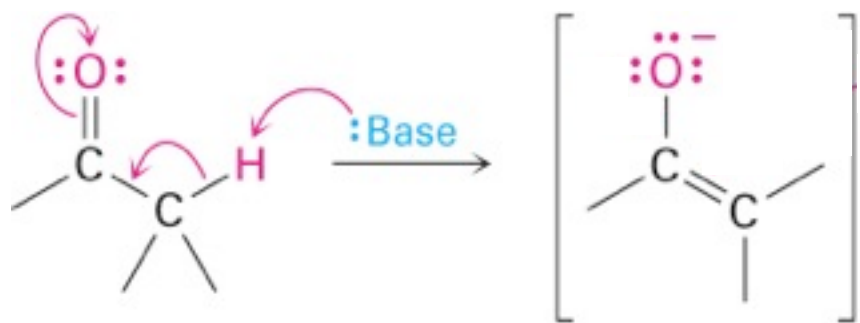
2 An electron pair from oxygen expels chloride ion and yields the substitution product, an ester.

2 ↓



Alpha Hydrogens are Acidic

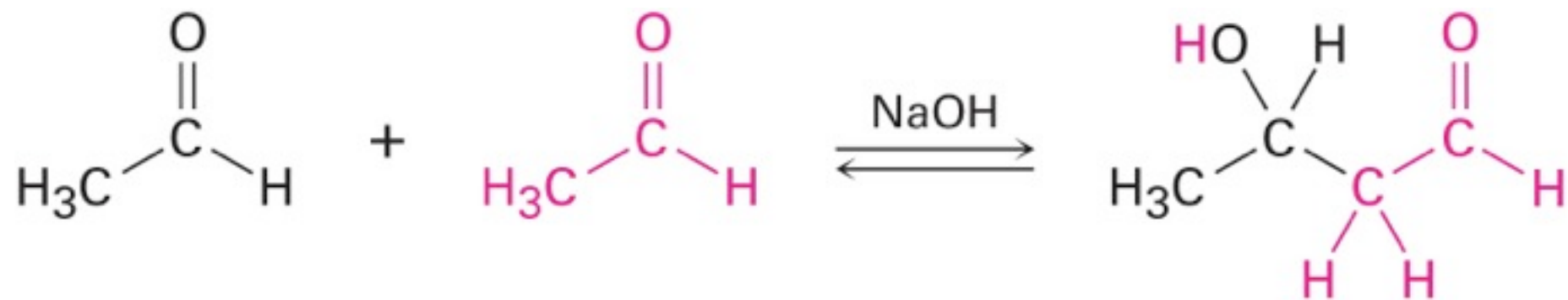




A carbonyl compound

An enolate ion

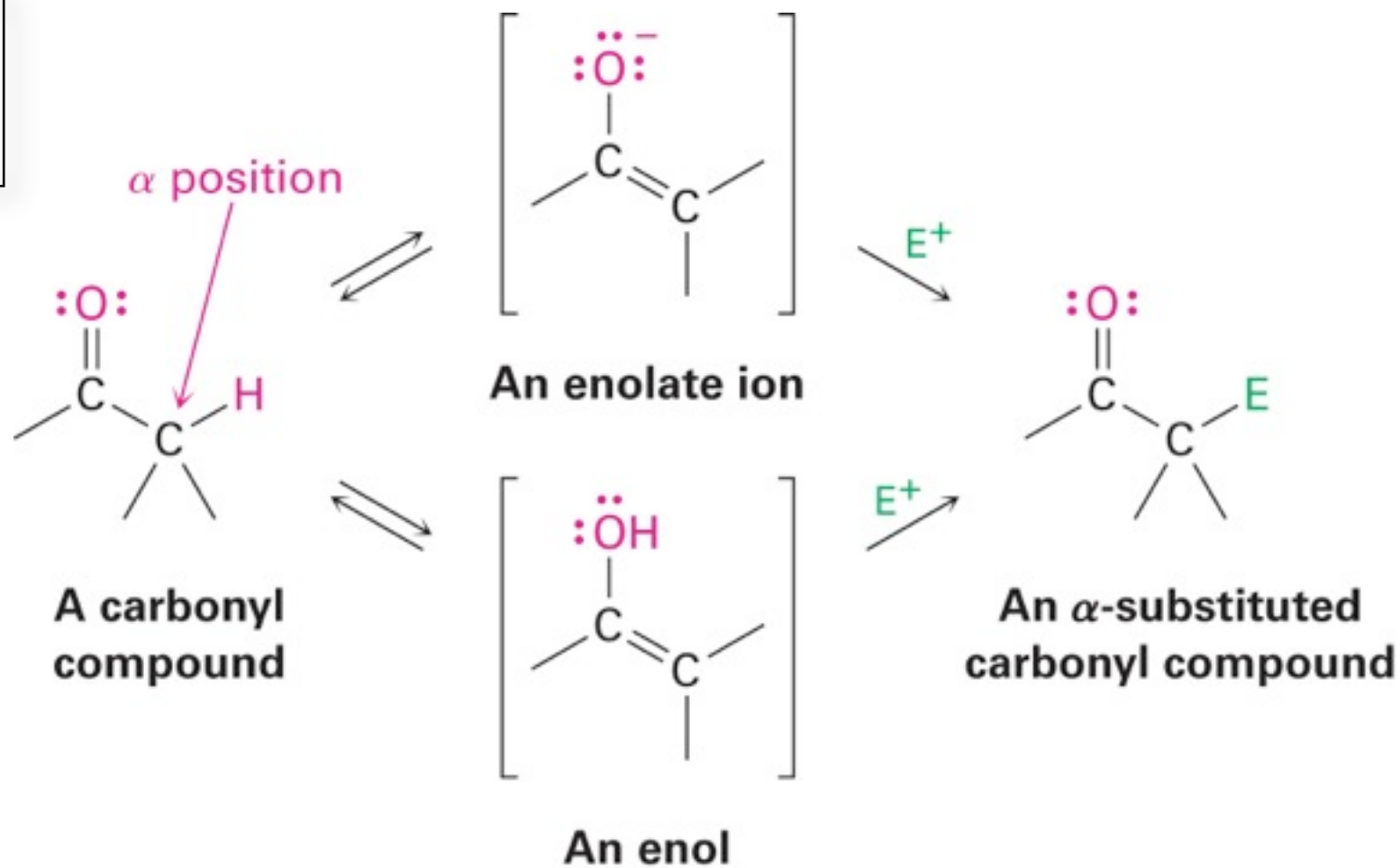
in almost all cases,
enolates are carbon
nucleophiles



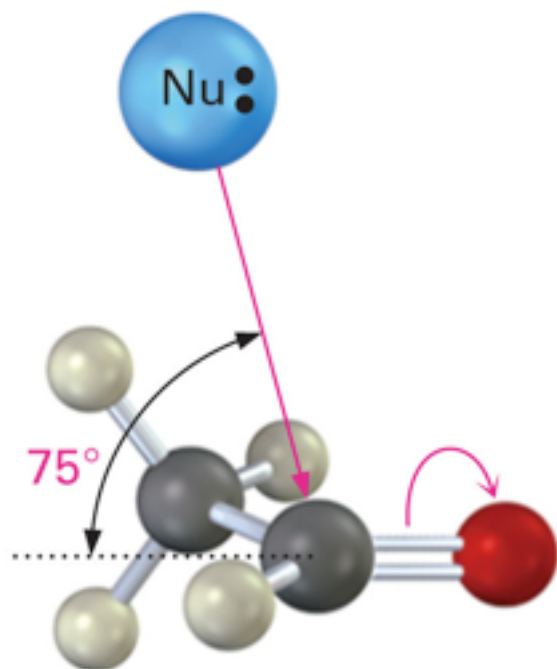
Two acetaldehydes

Aldol

Alpha Substitution



(a)



(b)

